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Next

|     | (1)                                                                                   | (2)                                                                               | (3)                                                                          |
|-----|---------------------------------------------------------------------------------------|-----------------------------------------------------------------------------------|------------------------------------------------------------------------------|
| (a) | <p>C</p> <p>2s 2p</p> <p>↑ ↑ ↑ ↑</p> <p>híbridos <math>sp^3</math></p> <p>↑ ↑ ↑ ↑</p> | <p>B</p> <p>2s 2p</p> <p>↑ ↑ ↑</p> <p>híbridos <math>sp^2</math></p> <p>↑ ↑ ↑</p> | <p>Be</p> <p>2s 2p</p> <p>↑ ↑</p> <p>híbridos <math>sp</math></p> <p>↑ ↑</p> |
| (b) | <p>orbital híbrido</p> <p>C</p>                                                       | <p>B</p>                                                                          | <p>Be</p>                                                                    |
| (c) | <p>Tetraédrica</p> <p>CH<sub>4</sub></p>                                              | <p>Triangular plana</p> <p>BF<sub>3</sub></p>                                     | <p>Lineal</p> <p>BeF<sub>2</sub></p>                                         |

## Tabla periódica de los elementos

clasifica, organiza y distribuye los distintos elementos químicos, conforme a sus propiedades y características

### Tabla periódica de los elementos

Legend:

- Metales alcalinos
- Metales alcalinotérreos
- Metales de transición
- Gases nobles
- Lantánidos
- Actínidos

### Tabla Periódica de los Elementos

Diagrama de la Tabla Periódica de los Elementos con anotaciones de tendencias y configuración electrónica.

**Anotaciones de Tendencia:**

- AE (Energía de Ionización):** Aumenta hacia la derecha (AE) y disminuye hacia la izquierda (RA).
- RA (Radio Atómico):** Aumenta hacia la izquierda (RA) y disminuye hacia la derecha (AE).
- Configuración Electrónica:** Se muestra la configuración para los primeros tres elementos:  $1s^1$  (H),  $1s^2$  (He),  $1s^2 2s^1$  (Li).

**Tabla Periódica:**

|          |          |          |           |           |           |           |           |           |           |           |           |           |           |           |           |           |           |          |
|----------|----------|----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|-----------|----------|
| 1<br>H   |          |          |           |           |           |           |           |           |           |           |           |           |           |           |           |           | 2<br>He   |          |
| 3<br>Li  | 4<br>Be  |          |           |           |           |           |           |           |           |           |           |           |           |           |           |           |           | 10<br>Ne |
| 11<br>Na | 12<br>Mg |          |           |           |           |           |           |           |           |           |           |           |           |           |           |           |           | 18<br>Ar |
| 19<br>K  | 20<br>Ca | 21<br>Sc | 22<br>Ti  | 23<br>V   | 24<br>Cr  | 25<br>Mn  | 26<br>Fe  | 27<br>Co  | 28<br>Ni  | 29<br>Cu  | 30<br>Zn  | 31<br>Ga  | 32<br>Ge  | 33<br>As  | 34<br>Se  | 35<br>Br  | 36<br>Kr  |          |
| 37<br>Rb | 38<br>Sr | 39<br>Y  | 40<br>Zr  | 41<br>Nb  | 42<br>Mo  | 43<br>Tc  | 44<br>Ru  | 45<br>Rh  | 46<br>Pd  | 47<br>Ag  | 48<br>Cd  | 49<br>In  | 50<br>Sn  | 51<br>Sb  | 52<br>Te  | 53<br>I   | 54<br>Xe  |          |
| 55<br>Cs | 56<br>Ba |          | 72<br>Hf  | 73<br>Ta  | 74<br>W   | 75<br>Re  | 76<br>Os  | 77<br>Ir  | 78<br>Pt  | 79<br>Au  | 80<br>Hg  | 81<br>Tl  | 82<br>Pb  | 83<br>Bi  | 84<br>Po  | 85<br>At  | 86<br>Rn  |          |
| 87<br>Fr | 88<br>Ra |          | 104<br>Rf | 105<br>Db | 106<br>Sg | 107<br>Bh | 108<br>Hs | 109<br>Mt | 110<br>Ds | 111<br>Rg | 112<br>Cn | 113<br>Nh | 114<br>Fl | 115<br>Mc | 116<br>Lv | 117<br>Ts | 118<br>Og |          |
| 57<br>La | 58<br>Ce | 59<br>Pr | 60<br>Nd  | 61<br>Pm  | 62<br>Sm  | 63<br>Eu  | 64<br>Gd  | 65<br>Tb  | 66<br>Dy  | 67<br>Ho  | 68<br>Er  | 69<br>Tm  | 70<br>Yb  | 71<br>Lu  |           |           |           |          |
| 89<br>Ac | 90<br>Th | 91<br>Pa | 92<br>U   | 93<br>Np  | 94<br>Pu  | 95<br>Am  | 96<br>Cm  | 97<br>Bk  | 98<br>Cf  | 99<br>Es  | 100<br>Fm | 101<br>Md | 102<br>No | 103<br>Lr |           |           |           |          |

[illegible]

CONFIGURACION ELECTRONICA DE LOS ELEMENTOS

| Número Atómico | Nombre    | Símbolo | Configuración Electrónica                                                                                                                                                         |
|----------------|-----------|---------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Z = 1          | Hidrógeno | H:      | 1s <sup>1</sup>                                                                                                                                                                   |
| Z = 2          | Helio     | He:     | 1s <sup>2</sup>                                                                                                                                                                   |
| Z = 3          | Litio     | Li:     | 1s <sup>2</sup> 2s <sup>1</sup>                                                                                                                                                   |
| Z = 4          | Berilio   | Be:     | 1s <sup>2</sup> 2s <sup>2</sup>                                                                                                                                                   |
| Z = 5          | Boro      | B:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>1</sup>                                                                                                                                   |
| Z = 6          | Carbono   | C:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>2</sup>                                                                                                                                   |
| Z = 7          | Nitrógeno | N:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup>                                                                                                                                   |
| Z = 8          | Oxígeno   | O:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>4</sup>                                                                                                                                   |
| Z = 9          | Flúor     | F:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>5</sup>                                                                                                                                   |
| Z = 10         | Neón      | Ne:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup>                                                                                                                                   |
| Z = 11         | Sodio     | Na:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>1</sup>                                                                                                                   |
| Z = 12         | Magnesio  | Mg:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup>                                                                                                                   |
| Z = 13         | Aluminio  | Al:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>1</sup>                                                                                                   |
| Z = 14         | Silicio   | Si:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>2</sup>                                                                                                   |
| Z = 15         | Fósforo   | P:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup>                                                                                                   |
| Z = 16         | Azufre    | S:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>4</sup>                                                                                                   |
| Z = 17         | Cloro     | Cl:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>5</sup>                                                                                                   |
| Z = 18         | Argón     | Ar:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup>                                                                                                   |
| Z = 19         | Potasio   | K:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>4</sup> 4s <sup>1</sup>                                                                                   |
| Z = 20         | Calcio    | Ca:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>4</sup> 4s <sup>2</sup>                                                                                   |
| Z = 21         | Escandio  | Sc:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>1</sup> 4s <sup>2</sup>                                                                   |
| Z = 21         | Escandio  | Sc:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>1</sup> 4s <sup>2</sup>                                                                   |
| Z = 22         | Titanio   | Ti:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>2</sup> 4s <sup>2</sup>                                                                   |
| Z = 23         | Vanadio   | V:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>3</sup> 4s <sup>2</sup>                                                                   |
| Z = 24         | Cromo     | Cr:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>4</sup> 4s <sup>1</sup>                                                                   |
| Z = 25         | Manganeso | Mn:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>4</sup> 4s <sup>1</sup>                                                                   |
| Z = 26         | Hierro    | Fe:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>4</sup> 4s <sup>2</sup>                                                                   |
| Z = 27         | Cobalto   | Co:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>4</sup> 4s <sup>2</sup>                                                                   |
| Z = 28         | Niquel    | Ni:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>4</sup> 4s <sup>2</sup>                                                                   |
| Z = 29         | Cobre     | Cu:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>1</sup>                                                                  |
| Z = 30         | Cinc      | Zn:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup>                                                                  |
| Z = 31         | Gallo     | Ga:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>1</sup>                                                  |
| Z = 32         | Germanio  | Ge:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup>                                                  |
| Z = 33         | Arsénico  | As:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup>                                                  |
| Z = 34         | Selenio   | Se:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup>                                                  |
| Z = 35         | Bromo     | Br:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup>                                                  |
| Z = 36         | Kriptón   | Kr:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup>                                                  |
| Z = 37         | Rubidio   | Rb:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 5s <sup>1</sup>                                  |
| Z = 38         | Estroncio | Sr:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 5s <sup>2</sup>                                  |
| Z = 39         | Itrio     | Y:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>1</sup> 5s <sup>2</sup>                  |
| Z = 40         | Circonio  | Zr:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>2</sup> 5s <sup>2</sup>                  |
| Z = 41         | Niobio    | Nb:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>3</sup> 5s <sup>1</sup>                  |
| Z = 42         | Molibdeno | Mo:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>5</sup> 5s <sup>1</sup>                  |
| Z = 43         | Tecnecio  | Tc:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>5</sup> 5s <sup>2</sup>                  |
| Z = 44         | Rutenio   | Ru:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 4d <sup>5</sup> 5s <sup>1</sup>                  |
| Z = 45         | Rodio     | Rh:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 4d <sup>5</sup> 5s <sup>1</sup>                  |
| Z = 46         | Paladio   | Pd:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 4d <sup>10</sup>                                 |
| Z = 47         | Plata     | Ag:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 4d <sup>10</sup> 5s <sup>1</sup>                 |
| Z = 48         | Cadmio    | Cd:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>5</sup> 4d <sup>10</sup> 5s <sup>2</sup>                 |
| Z = 49         | Indio     | In:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>1</sup> |
| Z = 50         | Estañó    | Sn:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>2</sup> |
| Z = 51         | Antimonio | Sb:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>3</sup> |
| Z = 52         | Telurio   | Te:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>4</sup> |
| Z = 53         | Yodo      | I:      | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup> 4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>5</sup> |
| Z = 54         | Xenón     | Xe:     | 1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>3</sup> 3s <sup>2</sup> 3p <sup>3</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup> 4d <sup>10</sup> 5s <sup>2</sup> 5p <sup>6</sup> |
| Z = 55         | Cesio     | Cs:     | [Xe]6s <sup>1</sup>                                                                                                                                                               |
| Z = 56         | Bario     | Ba:     | [Xe]6s <sup>2</sup>                                                                                                                                                               |
| Z = 57         | Lantano   | La:     | [Xe]5d <sup>1</sup> 6s <sup>2</sup>                                                                                                                                               |
| Z = 57         | Lantano   | Hf:     | [Xe]4f <sup>14</sup> 5d <sup>2</sup> 6s <sup>2</sup>                                                                                                                              |
| Z = 72         | Hafnio    | Ta:     | [Xe]4f <sup>14</sup> 5d <sup>3</sup> 6s <sup>2</sup>                                                                                                                              |
| Z = 73         | Tántalo   | Ta:     | [Xe]4f <sup>14</sup> 5d <sup>3</sup> 6s <sup>2</sup>                                                                                                                              |
| Z = 74         | Volframio | W:      | [Xe]4f <sup>14</sup> 5d <sup>4</sup> 6s <sup>2</sup>                                                                                                                              |
| Z = 75         | Renio     | Re:     | [Xe]4f <sup>14</sup> 5d <sup>5</sup> 6s <sup>2</sup>                                                                                                                              |
| Z = 76         | Osmio     | Os:     | [Xe]4f <sup>14</sup> 5d <sup>6</sup> 6s <sup>2</sup>                                                                                                                              |
| Z = 77         | Iridio    | Ir:     | [Xe]4f <sup>14</sup> 5d <sup>7</sup> 6s <sup>2</sup>                                                                                                                              |
| Z = 78         | Platino   | Pt:     | [Xe]4f <sup>14</sup> 5d <sup>9</sup> 6s <sup>1</sup>                                                                                                                              |
| Z = 79         | Oro       | Au:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>1</sup>                                                                                                                             |
| Z = 80         | Mercurio  | Hg:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup>                                                                                                                             |
| Z = 81         | Talio     | Tl:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>1</sup>                                                                                                             |
| Z = 82         | Plomo     | Pb:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>2</sup>                                                                                                             |
| Z = 83         | Bismuto   | Bi:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>3</sup>                                                                                                             |
| Z = 84         | Polonio   | Po:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>4</sup>                                                                                                             |
| Z = 85         | Astato    | At:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>5</sup>                                                                                                             |
| Z = 86         | Radón     | Rn:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>6</sup>                                                                                                             |
| Z = 87         | Francio   | Fr:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>6</sup> 7s <sup>1</sup>                                                                                             |
| Z = 88         | Radio     | Ra:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>6</sup> 7s <sup>2</sup>                                                                                             |
| Z = 89         | Actinio   | Ac:     | [Xe]4f <sup>14</sup> 5d <sup>10</sup> 6s <sup>2</sup> 6p <sup>6</sup> 6d <sup>1</sup> 7s <sup>2</sup>                                                                             |

Afinidad electronica de los elementos de la tabla periodica. Definicion de afinidad electronica en la tabla periodica. Afinidad electronica periodo de la tabla periodica. Ejemplos de afinidad electronica en la tabla periodica. Propiedades de la tabla periodica afinidad electronica. Comportamiento de la afinidad electronica en la tabla periodica. Tabla de afinidad electronica de los elementos. Tabla periodica de afinidad electronica.

W., Larson. "Threshold Photodetachment of H-." Physical Review A, vol. Today, Cinco de Mayo is celebrated in the state of Puebla with speeches, parades, reenactments of the Battle of Puebla and even an international mole sauce festival. R., Murray, K. While many adults may enjoy Mexican beer and other drinks, a pinata can be a fun way for even kids to get in on the celebration. 731-750. 88, nº 7, 2002, pp. While sombreros and fake mustaches can come across as disrespectful to Mexican-Americans, buying food or other goods from Mexican-American businesses or learning about Mexican-American history can keep the spirit and intent of the holiday alive in a way that's fun for everyone. 8990-9000. L., Gerber, G., Ho, J., y Lineberger, W. Garcia de la Vega, y Hiroshi Tatewaki. "Fine Structure of Ca-, Sr-, Ba-, and Ra- from the Many-Body Theory Calculation." Physical Review A, vol. Davies, B. Gunion, R. While the Battle of Puebla didn't end the war with France, it was an important symbolic victory. 073003-1 to 073003-4. E., Ljungblad, U., y David J. Williams, W. "Systematic Study of the Stable States of C-, Si-, Ge-, and Sn- via Infrared Laser Spectroscopy." Physical Review A, vol. K. France was attempting to force a puppet monarchy on Mexico, and the victory of the smaller, mostly indigenous Mexican army against some of Europe's toughest soldiers inspired the people of Mexico. 1773-1783. 14, nº 3, 1995, pp. Scheer, M., Bilodeau, R. 6987-7002. 37, nº 9, 2004, pp. "Electron Spectrometry at the µeV Level and the Electron Affinities of Si and F." Journal of Physics B: Atomic, Molecular and Optical Physics, vol. However, you don't have to stick to these classic foods to have a great Cinco de Mayo if you don't want to. Photo Courtesy: Ruth Hartnup/WikiMedia Commons Traditional Mexican dishes can add extra flavor to the holiday. Hanstorp, D., y Gustafsson, M. L433-L437. K., y Hotopt, H. 1765-1773. Berkovits, D. Boaretto, E., Gbelberg, S., Heber, O., y Paul, M. S Pakhrin/WikiMedia Commons Cinco de Mayo is often celebrated in the United States with Mexican food and drinks, music, dancing and more. C., Brodie, C. Grihakin, 1961-1965. 2844-2856. "Laser Photodetachment Measurement of the Electron Affinity of Atomic Oxygen." Physical Review A, vol. 010501-1 to 010501-3. "A Study of the Low-Lying Electronic States of Fe2 and Co2 by Negative Ion Photoelectron Spectroscopy." Journal of Chemical Physics, vol. MORE FROM REFERENCE.COM Andersen, T., Haugen, H. "A Study of the Electronic Structures of Pd-2 and Pd2 by Photoelectron Spectroscopy." Journal of Chemical Physics, vol. J., Ingram, C. However, it's not a national holiday in Mexico, and most of the country doesn't observe it. 6104-6107. 35, nº 1, 2002, pp. Leopold, D. 53, nº 6, 1996, pp. "Measurement of the Electron Affinity of Cerium." Physical Review Letters, vol. M., K. ¿Qué significa un signo o en la afinidad electrónica? 3698-3701. 64, nº 2, 2001, pp. Mexican beer, including brands like Dos Equis and Corona, as well as liquor and mixed drinks, such as tequila and margaritas, are also popular. S. C., y Michael D. Calabrese, y Thompson, J. Morse. L483-L487. In the United States, Cinco de Mayo took off as a holiday when Mexican-American activists used it to celebrate Mexican and indigenous accomplishments during the civil rights movement in the 50s and 60s. Lykke, Andersen, T., y Lineberger, W. Blondel, C., Cacciani, P., Delsart, C., y Trainham, R. 34, nº 9, 2001, pp. 5783-5784. And of course, local parades, festivals, concerts or other events are always a good time. F., St. John Dixon-Warren, Lineberger, W. 43, nº 11, 1991, pp. 93, nº 10, 1990, pp. W., Carpenter, D. 97, nº 12, 1992, pp. Chicken flautas, salsa verde enchiladas or avocado quesadillas can replace ground beef tacos as the main course, or you can make mole poblano for a dish straight out of Puebla itself. 32, nº 3, 1985, pp. 106, nº 13, 1997, pp. C., y Haugen, H. Elote — grilled corn on the cob covered in mayonnaise, cotija cheese and spices — can add a street festival feel to a meal, while a bowl of pozole beef soup makes for a great appetizer. A., y G. "Binding Energies in Atomic Negative Ions. III." Journal of Physical and Chemical Reference Data, vol. 28, nº 6, 1999, pp. 1511-1533. Pegg, Lykke, K. 85, nº 1, 1986, pp. L281-L288. 55, nº 3, 1997, pp. ¿Qué es la afinidad electrónica y como varía en la tabla periódica? "An Experimental Investigation of the Atomic Europium Anion." Journal of Physics B: Atomic, Molecular and Optical Physics, vol. Vilkas, Yasuyuki Ishikawa, y Uzi Kaldor. 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However, it didn't become widespread until beer and liquor companies used the holiday to promote sales in the 80s.In the United States, Cinco de Mayo is often celebrated with tacos, guacamole, nachos and other Mexican-American foods. 40, nº 7, 1989, pp. While most Americans think of mariachi music when it comes to Cinco de Mayo, there's also grupera, banda, Norteño, Tejano and more to choose from, so don't be afraid to mix things up. "Yb- 6p1/2—Low-Lying Shape Resonance Rather than a Bound State." Journal of Physics B: Atomic, Molecular and Optical Physics, vol. You can even learn zapateado, the type of dance moves often performed to mariachi music, or watch an expert if you're feeling bashful. "Ultraviolet Photoelectron Spectroscopy of Molybdenum and Molybdenum Monoxide Anions." Journal of Chemical Physics, vol. A., y Haugen, H. 34, nº 14, 2001, pp. "Electron Affinity of Strontium." Physical Review Letters, vol. Neumark, D. "Binding Energies in Atomic Negative Ions. II." 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K., y Lineberger, W. 31, nº 8, 1998, pp. That's celebrated on September 16, the anniversary of the 1810 call by Miguel Hidalgo for independence from Spain, and is comparable to July 4th and the signing of the American Declaration of Independence. Photo Courtesy: S Pakhrin/WikiMedia Commons Cinco de Mayo is instead the celebration of the victory of a Mexican Army over invading French forces at the Battle of Puebla on May 5th, 1862. "Measurement of the Electron Affinity of Thulium." Physical Review A, vol. ¿Cuál es el elemento de mayor radio atomico? C. 95, nº 7, 1991, pp. "Extrapolated Intermediate Hamiltonian Coupled-Cluster Approach: Theory and Pilot Application to Electron Affinities of Alkali Atoms." Journal of Chemical Physics, vol. "Laser Photodetachment Electron Spectrometry of Ga-" Journal of Physics B: Atomic, Molecular and Optical Physics, vol. "Photoelectron Spectroscopy of Metal Cluster Anions: Cu-n, Ag-n, and Au-n." Journal of Chemical Physics, vol. Bilodeau, R. 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